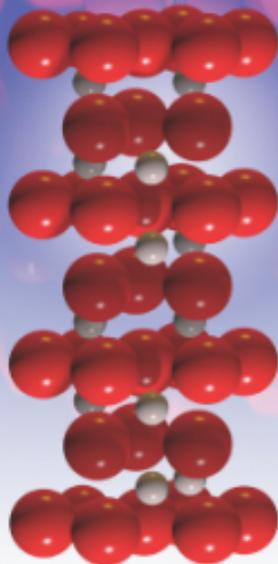


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Characteristics of Selected Elements

<i>Element</i>	<i>Symbol</i>	<i>Atomic Number</i>	<i>Atomic Weight (amu)</i>	<i>Density Solid, 20°C (g/cm³)</i>
Aluminum	Al	13	26.98	2.71
Argon	Ar	18	39.95	—
Barium	Ba	56	137.33	3.5
Beryllium	Be	4	9.012	1.85
Boron	B	5	10.81	2.34
Bromine	Br	35	79.90	—
Cadmium	Cd	48	112.41	8.65
Calcium	Ca	20	40.08	1.55
Carbon	C	6	12.011	2.25
Cesium	Cs	55	132.91	1.87
Chlorine	Cl	17	35.45	—
Chromium	Cr	24	52.00	7.19
Cobalt	Co	27	58.93	8.9
Copper	Cu	29	63.55	8.94
Fluorine	F	9	19.00	—
Gallium	Ga	31	69.72	5.90
Germanium	Ge	32	72.64	5.32
Gold	Au	79	196.97	19.32

Values of Selected Physical Constants

<i>Quantity</i>	<i>Symbol</i>
Avogadro's number	N_A
Boltzmann's constant	k
Bohr magneton	μ_B
Electron charge	e
Electron mass	—
Gas constant	R
Permeability of a vacuum	μ_0
Permittivity of a vacuum	ϵ_0
Planck's constant	h
Velocity of light in a vacuum	c

^a In cgs-emu units.

^b In cgs-esu units.



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Front Cover: Depiction of a unit cell for α -aluminum oxide (left) and iron sulfide (right) respectively.

Back Cover: (Top) Representation of a unit cell for iron sulfide, where iron and sulfur atoms are represented by yellow and blue spheres, respectively. (Bottom) Depiction of a unit cell for wurtzite, which is a form of zinc sulfide, where zinc and sulfur atoms are represented by yellow and blue spheres, respectively.

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